PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



SHED KINDER THE PATENT (YAODER ATTON TREATY (PCT)

	Institution 5 :	1	1	(11) International Publication Number: WO 95/0250
B32B 3/00			A1	(43) International Publication Date: 26 January 1995 (26.01.9
II) International Applica	ion Numbers	PCT/US9	14/0777	(81) Designated States: CA. IP, ER, European patent (AT, BE, C
(2) International Filing D		ш 1994 (2	8.06.9	DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE
		•		Published
30) Priority Data: 06/091,700	14 July 1999 (14.D	7.93)	U	With International search report.
71) Applicant: CHOMER Wobum, MA 0188	ICS, INC. [US/US]; (US).	77 Drago	o Com	
72) Inventor: LITMAN, (MA 01701 (US).	lary, J.; 472 Potter R	toed, Pour	inghan	•
74) Agents: HUBBARD, J Patent Dept., 55 H (US).	ohn, D. et al.; W.R. G layden Avenue, Lexi	irace & Co ngton, Mi	-Cons	
Thering Alberta (T	Ro) is used as a ther	melly con	daetive	filler in thermally conductive materials. The TiB2 provides improv
Titanium diboride (Ti hermal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrica
Titanium diboride (Ti hermal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti hermal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti termal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti termal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti termal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti termal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti termal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti termal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti hermal conductivity in co- conductive fillers especially	mandan to convent	ional them	mally i	onduction, electrically inculative or thermally conductive electrical
Titanium diboride (Ti hermal conductivity in co- conductive fillers especially	meanism to convent at low application y	ional them	The m	onductive, electrically insulative or thermally conductive electrical atorials into which the TiB ₂ filler is inscriporated includes alastome
Titanium diboride (Ti hermal conductivity in co- conductive fillers especially	meanism to convent at low application y	ional them	The m	onduction, electrically inculative or thermally conductive electrical
Thenium diborido (Ti hermal conductivity in co conductive fillers especially lims and tapes.	meanism to convent at low application g	ional them	The m	onductive, electrically insulative or thermally conductive electrical atorials into which the TiB ₂ filler is inscriporated includes alastome
hermal conductivity in co- conductive fillers especially lims and tapes.	meanism to convent at low application y	ional them	The m	
Titenium diboride (Ti hermal conductivity in co conductive fillers especially lims and tapes.	meanism to convent at low application g	ional them	The m	orductive, electrically insulative or thermally conductive electrical attributes which the TiB ₂ filler is incorporated includes electrical attributes and the tributes are supported included electrical attributes and the tributes are supported includes and the tributes are supported includes alantomer attributes and tributes are supported includes alantomer attributes are supported in tributes and the tributes are supported in tributes and tributes are supported and tributes are supported in tributes and tributes are supported and tributes are supported and tributes are supported at the tribu
Titanium diboride (Ti ermai conductivity in co andactive fillers especially ims and tapes.	meanism to convent at low application g	ional them	The m	onductive, electrically ingulative or thermally confuctive electrical attacks into which the TiB2 filler is incorporated includes electrons also attacks and the the TiB2 filler is incorporated includes electrons attacks.

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

	AT	Austia	GB	United Kingdom	8438	Markenia
	AU	Australia '	GE	Georgia	PEM	Malabi
	202	Barbados	GN	Culma	NZ	Niger
	82	Beleiten	GE.	Greece	ROL	Noteringe
١	57	Burides Faco	EU	Honesey	KO	Normay
ľ	RG	Bolgaria.	TE:	Entended.	NZ	New Zealand
	27	Beniz	<u> </u>	Italy	PL	Polund
	89	Breedi	JP	Japan	PT	Portugal
	BY	Selecus	KE	Kon/s.	RO	Romania
	CA	Сапиба	KO	Kyrgystan	RIT	Bussian Pederation
	<u></u>	Central African Republic	EP	Dunocratic People's Republic	60	Street
	œ	Compo	_	of Koma	38	Cwater
	Œ	Switzerland	ER	Republic of Korea	88	Slovenia .
	æ	Côta d'Ivoire	E2	Karabharan	88	Slovaida
	ä	Carremon	ū	Licchtonatelo	an	Sotreti
	ä	Chima		Sri Lamba	110	Chud
	ČŠ.	Crachodondda	<u> </u>	Luxanboura	TG	Taga
	ČŽ	Cruch Republic	LV	Labrie	IJ	Talikhstan
	OB	Germany	MC	Mogaço	n	Tripland and Tobaso
	DK	Decement	3420	Republic of Maldova	DA	Ulcraine
		Specia	MG	Madegacer	CS.	United States of America
	7	Pintand	ML	MAB	UZ	Usbekistan
	FR	Practic	MEN	Mongotta	VX	Viet Nam.
			241		764	7

PCT/US94/07772 WO 95/02504

THERMALLY COMDUCTIVE MATERIALS CONTAINING TITANIUM DIBORIDE FILLER

This invention relates to thermally conductive materials, such as elastomers, films, tapes, and compounds such as adhesives and greases. Moreover, it relates to the use of titanium diboride as a thermally conductive filler in such materials.

BACKGROUND OF THE INVENTION

Thermally conductive materials include elastomeric materials such as CHO-THERM® thermal pads, films, such as KAPTONO MT thermally conductive polyimide film, tapes, such THERMATTACH™ pressure sensitive acrylic tapes, and adhesives and greases, such as silicone grease, all of which contain a thermally conductive filler in a resin binder. They are primarily used in electronic applications where good thermal conduction is needed. For example, a thermally conductive material may be used as an interface between a heat source, such as semiconductor or other electronic component and a metal heat sink.

Many electronic designs and applications are limited by the ability to dissipate thermal energy generated during the operation of the electronic components. Many electronic components, especially semiconductor components, are prone to breakdown at high temperatures. Thus, the ability to dissipate heat is a limiting factor on the performance of the component.

Thermally conductive materials have been based upon the use of a binder, preferably a resin binder in particular a silicone, thermoplastic rubber, urethane or acrylic, into which one or more thermally conductive fillers have been distributed.

These fillers have typically been one of two major types: thermally conductive, electrically insulative or thermally conductive, electrically conductive fillers.

Aluminum cxide, magnesium oxide, zinc oxide, boron nitride and aluminum nitride are the most often cited types of thermally conductive, electrically insulative

PCT/US94/07772 WO 95/02504

fillers used in thermal products. Aluminum, copper, graphite, nickel and silver are commonly cited examples of thermally conductive, electrically conductive fillers.

In using such materials, it is known that by applying pressure between the heat source and the heat sink, one is able to obtain higher apparent thermal conductivity from the thermal product which is located between the heat source and heat sink. Typically, pressures of up to 300 psi are used in order to obtain the maximum apparent thermal conductivity possible from the product.

What is desired is a thermally conductive filler with higher thermal properties. Moreover, it is desired to have a thermally conductive material which has satisfactory thermal properties at low or no applied pressures.

The present invention provides such a material.

SUMMARY OF THE INVENTION

The invention provides filled thermally conductive material of improved apparent thermal conductivity using titanium diboride as the filler.

In one aspect, the invention encompasses a thermally conductive filled elastomer composition comprising an elastomer such as silicone rubber, thermoplastic rubber or polyurethane, and titanium diboride as the thermally conductive filler.

In another aspect, the invention encompasses a thermally conductive filled plastic film comprising a composition of a plastic resin and titanium diboride as a filler.

In a further aspect, the invention relates to a pressure sensitive, thermally conductive tape composition of a pressure sensitive silicone, urethane or acrylic resin and titanium diboride filler.

In an additional aspect, the invention relates to a thermally conductive adhesive or grease formed of a resin binder and titanium diboride filler.

PCT/US94/07772

DETAILED DESCRIPTION OF THE INVENTION

The compositions of the invention generally comprise a resin binder and titanium diboride as a thermally conductive filler. The compositions can be formed into various configurations such as elastomeric pads, films, tapes, compounds, adhesives and gresses.

Titanium diboride (TiB2) is a thermally conductive material. It is not electrically insulative, as is alumina or boron nitride. However, its electrical conductivity is at least a magnitude of order below the traditional metal fillers, such as aluminum copper, nickel or silver. As such it is useful for most thermal applications, especially those applications where electrical insulation is not an absolute necessity.

Moreover, the literature has described titanium diboride as having poorer thermal properties than the traditional fillers such as alumina and boron mitride. Applicant has found that this material when in particulate form and distributed throughout a resin matrix has high levels of apparent thermal conductivity. It has been found that the level of apparent thermal conductivity of a resin matrix containing titanium diboride as a thermally conductive filler is better than a resin matrix made using alumina, boron nitride, aluminum, copper, or silicon carbide under applied pressure or at no applied pressure at all.

The resin binder to which the filler may be added includes silicones, urethanes, thermoplastic rubbers, acrylics and the like.

Silicone elastomers are preferably formed from a silicone gum which is crosslinked using a catalyst. An example of a suitable silicone gum is sold under the name "Silastic 4-2765" by Dow Corning, Inc. A peroxide catalyst: 2,5-dimethyl 2,5-bis (t-butyl peroxy) hexane

8602860115

PCT/US94/07772

50% on CaCO3 sold by R. T. Vanderbilt as Varox® DBPH-50 is an example of a suitable catalyst.

Proferred thermoplastic rubbers are typically thermoplastic blockpolymers such as Kraton® G-1657-MX (a styrene-ethylene-butylene-styrene block copolymer having a styrene/rubber ratio of 13/87) sold by Shell Chemical Co.

Preferred crosslinkable block copolymers are thermoplastic rubbers such as Kraton® G-1901-X and Kraton® RP-6501 sold by Shell Chemical Co. These rubbers are styrene/clefin based with functional groups. G-1901-X has carboxyl groups whereas RP-6501 has ethoxysilanol groups. In order to cause crosslinking, a crosslinking agent and a crosslinking catalyst are preferably combined with the orosslinkable copolymer. A preferred crosslinking agent is Cymel® 303 (hexamethoxymethylmelamine) sold by American Cyanamid. A preferred crosslinking catalyst for G-1901-X is Cycato 600 (aromatic sulfonic acid) also sold by American Cyanamid. For RP-6501, dibutyl tin dilaurate is a preferred catalyst.

Preferred urethanes are polyurathane resing which are nonfoaming and capable of being crosslinked at elevated temperatures. Such systems are well known, such as from U.S. Patent 4,869,954 which is incorporated by reference in its entirety.

Preferred resins for films include polyolefins such as polyethylene, polyimides, polyamides, polyesters and PREK. Well known films particularly useful in the present invention are MYLARO polyester films and KAPTONO polyimide films. As these films, without a thermally conductive filler, exhibit poor thermal conductivities, the additional of titanium diboride alone or in combination with one or more thermally conductive fillers, especially boron nitride or aluminum oxide, increases the thermal properties of the film dramatically, while causing little or not effect to the film's physical properties.

WO 95/02504 PCT/US94/07772

When the thermally conductive material is in the form of a tape or an adhesive, it is preferably formed of a pressure sensitive adhesive material, such as a silicone, urethane or an acrylic adhesive resin and titanium diboride, alone or in combination with one or more additional thermally conductive fillers. Such resins are well known and commercially available. <u>Sac</u> U.S. 5,213,868 which is incorporated by reference in its entirety.

The tape is typically formed by costing the composition containing the filler and resin onto a releasable surface and dried or cured to form a form stable material. If desired, a releasable coversheet may be applied to one or both sides of the tape to keep it from prematurely sticking to a surface, to allow it to be rolled up on itself and/or to keep it from picking up dirt, dust and often debris which could interfere with the thermal capabilities of the tape in use.

When the thermally conductive material is in the form of a grease, it is preferably formed of an uncrosslinked silicone oil containing titanium diboride as a thermally conductive filler.

If desired in the elastomer or tape configuration, one may incorporate one or more layers of a support material to increase the toughness, resistance to elongation and tearing and other mistreatment of the thermally conductive material. The incorporation of support materials is particularly desired where mechanical strength needs to be improved. Any tearing or penetration of the thermally conductive material allow for the formation of a pocket of air, which as described above is poor thermal conductor. These air pockets can lead to localized hot spots and an overall decline in thermal performance.

The support material should not adversely interfere with the thermal conductivity of the thermally conductive material. Preferably, the support material is itself thermally conductive.

PCT/US94/07772

Examples of reinforcing materials useful in the present invention include, but are not limited, to glass fiber, mesh or cloth, plastic fiber, mesh cloth or films and metal fiber, mesh, cloth or foils.

The glass cloth may be woven or unwoven.

The plastic support material is preferably a mesh material, or a film. If a film, it may be solid or foraminous. Examples of suitable plastics include nylons, polyesters, polamides, polyimides, polyethylenes, or PEKK. Well-known plastic films particularly useful in the present invention are MYLAR® polyester films and KAPTON® polyimide films. As plastic films generally exhibit poor thermal conductivity properties, it is desirable to use a film which contains a thermally conductive filler. One such filler film is KAPTON MT®, a polyimide film which contains either an aluminum oxide or boron nitride filler. This material exhibits twice the thermal conductivity of an equivalent unfilled film. Alternatively, one may use a very thin film layer so as to minimize its effects on the thermal transfer.

Metal mesh cloths or foils are the preferred support material due to their high thermal conductivity. Examples of preferred metals useful as a mesh cloth or a foil, include but are not limited to, aluminum, copper, silver, iron, tinned copper, MONEL or FERREXO metal. Regardless of the support material used, it should be as thin as practicable while still providing the desired support.

The support materials may be embedded in the thermally conductive material or coated, calendared or attached to one surface of the thermally conductive material. Preferably, the support material and thermally conductive material are formed into a laminate of three or more layers with the support layer being the center layer and the thermally conductive material forming the outer layers.

WO 95/02594 PCT/US94/07772

The thermally conductive filler comprises titanium diboride (TiB₂). Other thermally conductive fillers may be used in conjunction with the titanium diboride, however, compositions having titanium diboride as the only thermally conductive filler are preferred. The filler preferably has an average diameter of about 2-50 microns (smaller diameters in this range are more preferred) and may be in any form such as particles, spheres and/or platelets. If desired, one may use a range of particle sizes to achieve higher loadings. The filler may be obtained commercially or may be formed by any known process.

The proportion of TiB₂ filler in the filled composition may vary depending on factors such as the desired thermal conductivity, the degree of electrical isolation and the degrees of elasticity desired, etc. Generally, the thermal conductivity increases with the proportion of filler in the composition. Preferably, the composition contains at least about 60 wt.\filler more preferably at least about 70 wt.\filler. Alternatively, the titanium diboride constitutes from about 10 to about 60 volume \forall of the composition, preferably from about 40 to about 70 volume \forall. The balance of the composition is formed by the resin and any auxiliary materials (e.g. catalysts, residual solvents, plasticizers, other fillers, etc.) Any appropriate conventional additives may be incorporated into the composition.

The thermal materials may be formed by any conventional method in any conventional shape. The methods all involve the formation of an initial mixture of resin and TiB2 filler, which then can be formed into a sheet film or tape or placed in a mold, dried and/or cured.

One method of forming the material is to combine the resin binder with the selected filler or fillers and

PCT/US94/07772

thoroughly mix the ingredients while slowly adding a solvent until a liquid having a smooth texture is achieved. The material is then cast onto a release sheet such as a piece of glass, MYLAR® film or coated paper, or on to a support layer and heated to drive off the solvent and form the thermally conductive material.

An alternative method is to thoroughly mix the ingredients together with a sufficient amount of solvent to obtain a thin liquid. The liquid can then be sprayed or coated onto a surface such as a release sheet or a reinforcing material such as a glass fiber cloth, a KAPTONO film or metal foil and heated to cure. Additionally, the same liquid formulation may have a porous support material, such as a glass fiber mesh, dipped into it to form the desired coating.

A further method of forming a heat conductive material of the present invention is by extrusion. The resin is reduced molten and the filler and other ingredients, if any, are added to the melt stream and intimately mixed with the resin. The molten mixture is then extruded, as a flat sheet on a tube which can be expanded and slit to form a thin film, as is well known in the art.

Another method of forming the heat conductive material of the present invention is by molding. This is particularly useful when one wishes to form a substantially thick heat conductive layer or when one wishes to form a specially shaped heat conductive material such as an elastomeric pad. In molding the heat conductive material, the components are mixed with a solvent and poured into a prefabricated mold may coat the inside of the mold with a release coating before adding the components. The mold is then heated or otherwise subjected to an external energy field to form the molded shape. Instead of using a separate mold, it may be

WO 95/02504 PCT/US94/07772

desired to use a mold which allows the heat conductive material to be molded in place directly to one of the surfaces it will contact.

A preferred method is to form a laminated material of three or more layers in which the center layer is formed of a support layer discussed above, such as glass mash, plastic film or metal foil and an outer layer of the thermally conductive material is bonded on each side of the support layer and interpenetrating it so as to form a cohesive laminated material. The bonding may occur sequentially so that one side of the support layer is covered and cured and then the process is repeated on the opposite side. Preferably, the outer layers are applied to both sides simultaneously. If desired, all three layers may contain titanium diboride as the primary thermally conductive filler.

Where it is desired to have the thermal interface material provide electrical isolation as well as thermal conductivity, it is possible to do so, even with the electrically conductive titanium diboride filler when one forms a laminate of two outer layers containing titanium diboride and a center layer of thermally conductive, electrically insulative material such as a filled film, such as KAPTOMP MT film. In order to ensure that no arcing or flash around occurs between the two isolated conductive layers, the middle layer should be of a width and length that is greater than that of the two outer layers. Preferably, it has a width and length that is at least 5%, more preferably 10% greater than that of the outer layers. This prevents such aroing from occuring due to the long length of dielectric surface over which the electrical energy would have to travel. For example, a square piece of the product according to this embodiment would preferably have two layers containing titanium diboride filler of 1 inch by 1 inch and middle dielectric layer, such as of KAPTONO MT film of 1.1 inch by 1.1 inch.

PCT/US94/07772 WO 95/02504

The invention is further illustrated by the following example. The invention is not limited to the materials, process steps or results given in the examples below.

EXAMPLE

A thermally conductive material according to the present invention was made as follows:

- 1.) A mixture of 10 grams isopropyl alcohol and 40 grams of a solvent (toluene) was prepared and to which as added a silane coupling agent in an amount equal to 1% by weight of the thermal filler to be added. A small amount of deionized water was then added to prehydrolize the silane.
- 2.) Titanium diboride (43.20 grams) was added and mixed for about 30 minutes to form a slurry,
- 3.) The slurry was mixed with 36 grams of acrylic resin dissolved in toluene solvent. The slurry was then mixed under high shear conditions for 60 minutes and then degassed.
- 4.) The slurry was knife coated onto a release film in a thickness of 13 mils and dried in a hot air convection oven to drive off the solvent and then cured in the same oven at a temperature of 125% for 5 minutes. The coated release films were then laminated to both sides of 2 mil 1100 aluminum foil.

The product was then tested for thermal conductivity at levels of 0 psi and at 300 psi with the following results:

0 psi: 0.60 Units are W/m-K 300 psi: 0.72 Units are W/m-K

BI.q

PCT/US94/67772

As a comparison, the same product was made using 21.6 grams of boron nitride (21 volume % filler, same as the present example) and tested for thermal conductivity at 0 and 300 psi with the following results:

0 psi: 0.33 units are in W/m-K
300 psi: 0.61 units are in W/m-R

As can be clearly seen, the titanium diboride filler provided significantly higher thermal conductivity at 0 psi as compared to the boron nitride filler and better apparent thermal conductivity properties at 300 psi than the boron nitride filler example.

PCT/US94/07772 .

What I Claim:

- 1.) A thermally conductive material comprising a composition of a resin binder and thermally conductive filler of titanium diboride distributed throughout the resin binder.
- 2.) The material of Claim 1 wherein the composition is selected from the group consisting of elastomers, films, and tapes.
- 3.) The material of Claim 1 wherein the material is an elastomer selected from the group consisting of silicones, urethenes, and thermoplastic rubbers.
- 4.) The material of Claim 1 wherein material is a film is selected from the group consisting of mylons, polyimides, polyamides, polyesters, polyolefins, PEEK and blends thereof.
- 5.) The material of Claim 1 wherein the material is a tape selected from the group consisting of silicones, urethenes and acrylics.
- 6.) A thermal conductive tape comprising a pressure sensitive adhesive resin binder.
- 7.) The tape of Claim 6 wherein the resin is selected from the group consisting of acrylic resin urethane and silicone rubber.
- 8.) The tape of Claim 6 further comprising a support overlaying the adhesive layer and a second adhesive layer formed of a pressure sensitive resin binder and titanium diboride filler overlaying the support layer.
- 9.) The tape of Claim 6 wherein the titanium diboride is from about 10% to about 50% by volume of the layer.

WO 95/02504 PCT/DS94/07772

10.) The tape of Claim 6 further comprising one or more additional thermally conductive fillers selected from the group consisting alumina, boron nitride, magnesia, graphite, aluminum nitride and blends thereof.

- 11.) A thermally conductive elastomer comprising a composition formed of an elastomer selected from the group consisting of silicones, urethanes and thermoplastic rubbers, and a thermally conductive filler of titanium diboride.
- 12.) The elastomer of Claim 11 further comprising one or more additional thermally conductive fillers selected from the group consisting of alumina, boron nitride, magnesia, graphite, aluminum nitride and blends thereof.
- 13.) The elastomer of Claim 11 further comprising a support layer contained within the elastomer layer, the support layer being selected from the group consisting of plastic, metal and fiber glass materials.
- 14.) The elastomer of Claim 11 wherein the resin binder is a silicone rubber and the filler is present in an amount of from 10% to 60% volume % of the elastomer.
- 15.) A thermally conductive film comprising a film formed of a resin binder selected from the group consisting of polyolefins, nylons, polyimides, polyamides, polyesters. PEEK, and combinations thereof and titanium diboride as a filler.
- 16.) The film of Claim 15 further comprising one or more additional thermally conductive fillers selected from the group consisting of alumina, boron nitride, megnesia, graphite, aluminum nitride, zinc oxide and blends thereof.
- 17.) The film of Claim 15 wherein the resin binder is a polyimide.
- 18.) The film of Claim 15 wherein the filler is present in an amount of from about 10% to about 60% volume percent of the film.

WO 95/02504 PCT/US94/07772

first outer thermally conductive layer formed of a resin binder and a thermally conductive filler, a thermally conductive support layer overlaying the first outer layer, the support layer being formed as a film and comprised of a resin binder and thermally conductive filler, and a second outer layer formed on a surface of the support layer opposite that to which the support layer overlays the first outer layer, the second outer layer being formed of a resin binder and a thermally conductive filler, wherein the filler of the support layer is formed of titanium diboride and the filler of the first and second outer layers is selected from the group consisting of titanium diborate, alumina, magnesia, boron nitride, graphite, aluminum nitride and blends thereof.

20.) The composite of Claim 19 wherein the filler of the first and second outer layers is titanium diboride.

	INTERNATIONAL SEARCH REPOR	tr	International app PCT/US94/077					
A. CLASSIFICATION OF SUBJECT MATTER IPC(3) :28328 3/00 UE CL. :428/343, 355, 423.1, 425.9, 902, 240, 242. According to International Patent Chesification (IPC) or to both national classification and IPC								
	DS SEARCHED							
Minimum documentation searched (classification system followed by classification symbols) U.S.: 428/343, 355, 423.1, 425.9, 902, 240, 242								
Description reached other than minimum deconnectation to the extent that such decomments are included in the fields scarched NONE								
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) NONE								
C. DOCL	MENTS CONSIDERED TO BE RELEVANT							
Category*	Citation of document, with indication, where a	ppropriate, of the role	Vant pusages	Relevant to claim No.				
	US, A, 4,88,247 (ZWEBEN et column 7, lines 53-57 and column		1-20					
	US, A, 4,490,283 (KLEINER) 25 December 1984, column 5, 1-20 lines 62-63.							
				·				
Perther documents are listed in the continuation of Box C. See patent family ansex.								
_	ini cutagorius of cital documents: cuusi defizing the gaseral state of the art which is not considered past of particular palerapae	"I" jular dispussion data and not be vericelyle at the	published ofter the late conflict with the applies new underly but the last	emational filing data or priority allow but cloud to understand the audion				
E	o ciclosed investice senset be red to involve as investive map							
	ment which may three doubts on priority chila(s) or which to to articlash the publication date of quarter shadon or other all tensors (so specified) which influring to an east disclosure, was, whilehits or other			o delicant invention cannot be step when the document is a documents, such consideration a cut				
descenses published prior to the integrational filling date but inter that "A" document member of the some paint filmily the priority date obligated								
the priority data elegand Data of the actual completion of the international search Data of mailing of the international search report								
29 AUGUS	T 1994	2 1 SEP 1994						
Name and unit Commissions Box PCT Washington, I FaceignBo No.		Authorised officer E.ROLLINS BUI Telephone No. (7	PALOW 03) 302-2351	rave for				

E2.9